

THE SYNTHESIS OF 2,5-DIHYDROFURANS
 FROM α -ALLENIC ALCOHOLS

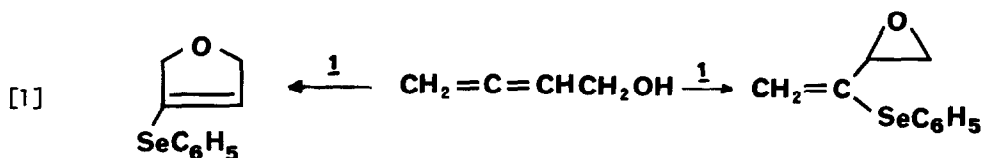
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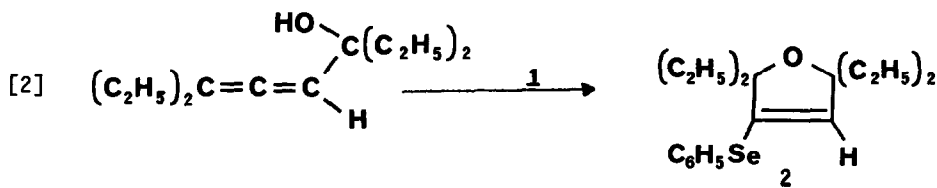
ABSTRACT: α -Allenic alcohols react rapidly with PhSeCl to produce derivatives of 3-phenylseleno-2,5-dihydrofuran in high yields.

The synthesis of heterocycles via the electrophilically-induced cyclization of unsaturated acids,¹ alcohols,² and phenols with areneselenenyl halides or derivatives thereof⁴ has recently received a great deal of attention by the groups of Clive and Nicolaou. In particular, Clive and coworkers have demonstrated the utility of the reaction of γ,δ unsaturated alcohols with benzeneselenenyl chloride, **1**, to produce ring-fused tetrahydrofurans^{2a}. While electrophilically-induced ring closures of this type have been known for many years⁵, the advantage of such a closure under these conditions arises from the facile nature of the process and the existence of the phenylseleno moiety as a potential synthon for further synthetic transformations^{6,7}.

The availability of diastereoisomeric α -allenic alcohols⁸, plus the known propensity for selenenyl halides to react with carbon-carbon double bonds in an *anti* stereospecific manner⁹, suggested to us the possibility of a facile synthetic route to configurationally well-defined 2,5-dihydrofurans. It must be noted, however, that conceptually there exist two distinct modes of cyclization if the phenylseleno moiety forms a new bond with the central carbon of the allenic system (equation [1]), which seems likely¹⁰.

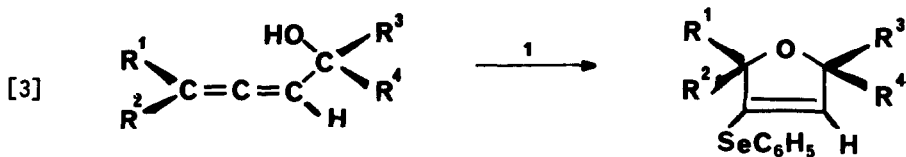


In order to establish the preferred pathway, we treated 3,6-diethyl-4,5-octadien-3-ol with 1 equiv. of **1** in methylene chloride at room temperature and found that it is rapidly transformed into 2,2,5,5-tetraethyl-3-phenylseleno-2,5-dihydrofuran, **2**, which was isolable in essentially quantitative yield. The ¹H NMR and ¹³C NMR spectra define the structure of **2**, and, as expected,



mass spectral analysis shows the production of fragments corresponding to $(M)^+$, $(M-C_2H_5)^+$, $(M-C_2H_5-C_6H_5Se)^+$, and $(C_6H_5Se)^+$.¹¹

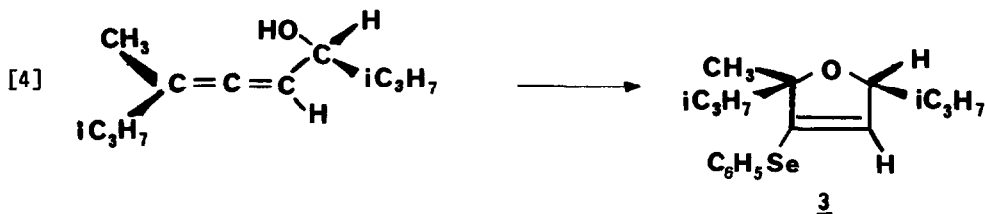
The generality of the ring closure is demonstrated by the following reactions:

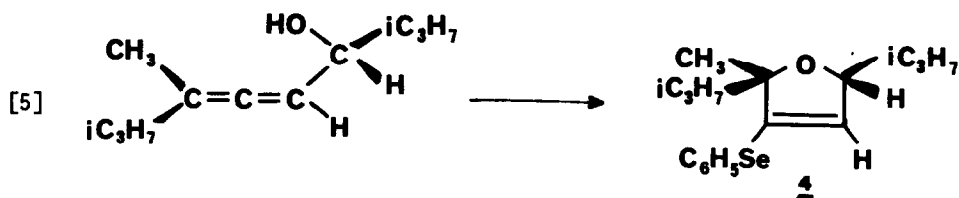


	R ¹	R ²	R ³	R ⁴	% Yield
a	CH ₃	CH ₃	CH ₃	C ₂ H ₅	90
b	CH ₃	CH ₃	CH ₃	iC ₃ H ₇	98 ⁺
c	CH ₃	CH ₃	CH ₃	tC ₄ H ₉	92
d	CH ₃	CH ₃	iC ₃ H ₇	iC ₃ H ₇	97
e	CH ₃	iC ₃ H ₇	CH ₃	iC ₃ H ₇	85
f	CH ₃	iC ₃ H ₇	iC ₃ H ₇	CH ₃	88
g	H	iC ₃ H ₇	H	iC ₃ H ₇	89
h	H	iC ₃ H ₇	iC ₃ H ₇	H	93
i	CH ₃	C ₂ H ₅	H	iC ₃ H ₇	79
j	CH ₃	C ₂ H ₅	iC ₃ H ₇	H	86
k	CH ₃	iC ₃ H ₇	H	iC ₃ H ₇	90
l	CH ₃	iC ₃ H ₇	iC ₃ H ₇	H	91
m	H	H	C ₂ H ₅	C ₂ H ₅	83
n	H	H	iC ₃ H ₇	iC ₃ H ₇	70
o	H	H	CH ₃	tC ₄ H ₉	98 ⁺
p	H	iC ₃ H ₇	C ₂ H ₅	C ₂ H ₅	95
q	H	iC ₃ H ₇	iC ₃ H ₇	iC ₃ H ₇	98 ⁺

Isolated yields are normally > 85%. The reactions involving terminal allenes [3n,n,o], are somewhat sluggish. They proceed faster and in better yields when 1 equiv. of triethylamine is added to react with the HCl which is given off.

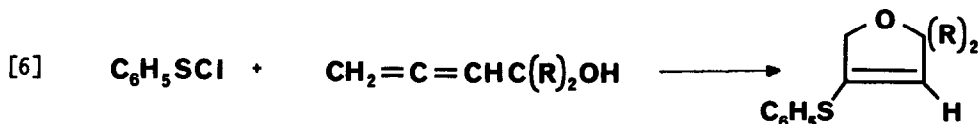
The stereochemistry of the ring closure is readily demonstrated by the reaction of 1 with diastereoisomeric α -allenic alcohols. For example, the reaction of 1 with (3-RS, 5-RS)-2,6,7-trimethyl-4,5-octadien-3-ol yields Z-2,5-diisopropyl-2-methyl-3-phenylseleno-2,5-dihydrofuran, 3, stereospecifically, whereas the corresponding (3-RS, 5-SR) allenic alcohol gives the analogous E-2,5-dihydrofuran (equations [4] and [5]).



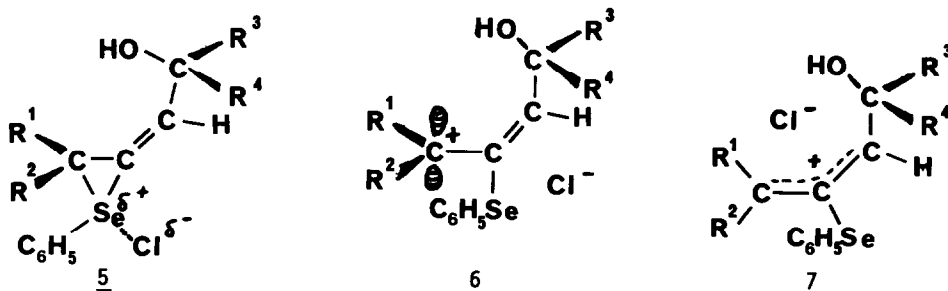


Assignment of 3 and 4 as the Z and E isomers respectively follows directly from their respective ^1H NMR spectra. Thus considering only the ring protons of the furan system, we observe for compound 3 a doublet at δ 5.34 1H $^3J_{4,5} = 1.50$ Hz and a doublet of doublets at δ 4.26 1H $^3J_{4,5} = 1.50$, $^3J_{5,6} = 7.50$ Hz, whereas for compound 4 we observe a doublet at δ 5.39 1H $^3J_{4,5} = 1.25$ Hz and a doublet of doublets at δ 4.26 1H $^3J_{4,5} = 1.25$, $^3J_{5,6} = 4.75$ Hz. The significant observation is the magnitude of the coupling constant $^3J_{5,6}$, where H_6 is the methine proton of the isopropyl group directly bonded to C_5 . An examination of molecular models reveals a pronounced steric interaction between the two isopropyl groups in compound 3 which gives rise to a conformer distribution favouring the *anti* conformer with respect to protons H_5 and H_6 . As this conformer is expected to be favoured more in compound 3 than in the case of 4 one will observe a larger value of $^3J_{5,6}$ for 3, as assigned. Similar results were observed for the other series of diastereoisomeric allenes studied.

These results have an analog in the reaction of 2,4-dinitrobenzenesulphenyl chloride with a partially resolved β -allenic alcohol¹² to give an optically active 3,4-dihydro-2H-pyran. In this respect we note our analogous results involving the reaction of benzenesulphenyl chloride with α -allenic alcohols to give 3-phenylthio-2,5-dihydrofurans in excellent yield.



In addition to the above, we observe second order kinetics, first-order in electrophile ($\text{C}_6\text{H}_5\text{SeCl}$ or $\text{C}_6\text{H}_5\text{SCl}$) and first-order in α -allenic alcohol, which establishes the composition of the rate determining transition state as containing one equivalent of each reactant. These data are indicative of a reaction hypersurface containing cyclic intermediate, 5, or a non-resonance stabilized carbonium ion, 6, which collapses to product before bond rotation to the resonance stabilized form 7 can occur.



A continuation of these studies towards the synthesis of 2,5-dihydrothiophenes and 2,5-dihydropyrroles is currently in progress.

REFERENCES & FOOTNOTES

1. (a) D.L.J. Clive and G. Chittattu, J.C.S. Chem. Comm. 484 (1977).
(b) K.C. Nicolaou and Z. Lysenko, J. Am. Chem. Soc. 99, 3185 (1977).
2. (a) D.L.J. Clive, G. Chittattu and C.K. Wong, Can. J. Chem. 55, 3894 (1977).
(b) K.C. Nicolaou and Z. Lysenko, Tetrahedron Letters 1257 (1977).
3. D.L.J. Clive, G. Chittattu, N.J. Curtis, W.A. Kiel, and C.K. Wong, J.C.S. Chem. Comm. 725 (1977).
4. (a) R.M. Scarborough Jr., A.B. Smith III, W.E. Barnette, and K.C. Nicolaou, J. Org. Chem. 44, 1742 (1979).
(b) K.C. Nicolaou, W.E. Barnette, and R.L. Magolds, J. Am. Chem. Soc. 100, 2567 (1978).
5. (a) D.L.H. Williams, Tetrahedron Letters 2001 (1967).
(b) T.L. Jacobs and R. Macomber, J. Org. Chem. 33, 2988 (1968).
6. (a) H.J. Reich, I.L. Reich, and J.M. Renga, J. Am. Chem. Soc. 95, 5813 (1973).
(b) K.B. Sharpless and R.F. Lauer, J. Am. Chem. Soc. 95, 2697 (1973).
(c) H.J. Reich and S.K. Shah, J. Am. Chem. Soc. 97, 3250 (1975).
(d) K.B. Sharpless, R.F. Lauer, and A.Y. Teranishi, J. Am. Chem. Soc. 95, 6137 (1973).
(e) P.A. Grieco and Y. Yokoyama, J. Am. Chem. Soc. 99, 5210 (1977).
(f) D.H.R. Barton, S.V. Ley, P.D. Magnus and M.N. Rosenfield, J.C.S. Perkin Trans. I, 567 (1977).
(g) J. Remion and A. Krief, Tetrahedron Letters 3743 (1976).
7. D.L.J. Clive, Aldrichimica Acta 11, 43 (1978).
8. (a) J.S. Cowie, P.D. Landor and S.R. Landor, J.C.S. Chem. Comm. 541 (1969).
(b) S.R. Landor, E.S. Pepper and J.P. Regan, J. Chem. Soc. (c) 189 (1967).
(c) P.D. Landor, S.R. Landor and E.S. Pepper, J. Chem. Soc. (c) 185 (1967).
9. (a) D.G. Garratt and G.H. Schmid, Can. J. Chem. 52, 3599 (1974).
(b) G.H. Schmid and D.G. Garratt, Tetrahedron Letters 3991 (1975).
(c) K.B. Sharpless and R.F. Lauer, J. Org. Chem. 39, 429 (1974).
(d) W. Jenny, Helv. Chim. Acta 36, 1278 (1953).
(e) D.G. Garratt, Can. J. Chem., 56, 2184 (1978).
10. (a) D.G. Garratt and P.L. Beaulieu, Can. J. Chem. 57, 119 (1979).
(b) D.G. Garratt, P.L. Beaulieu, and M.D. Ryan, Tetrahedron, in press.
11. Compound 2: ¹H NMR (δ, CDCl₃) 5.44 s (1H), 1.62 q (4H), 1.57 q (4H), 0.91 t (6H), 0.82 t (6H), 7.3 m (3H), 7.6 m (2H); ¹³C NMR (δ, CDCl₃) 94.0 s, 91.6 s, 31.6 t, 31.0 t, 8.7 q, 8.6 q, 128.1 d, 129.3 d, 134.7 d, 132.7 d, 131.6 s; colorless to pale yellow oil.
12. T.L. Jacobs, R. Macomber and D. Zucker, J. Am. Chem. Soc., 89, 7001 (1967).

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